

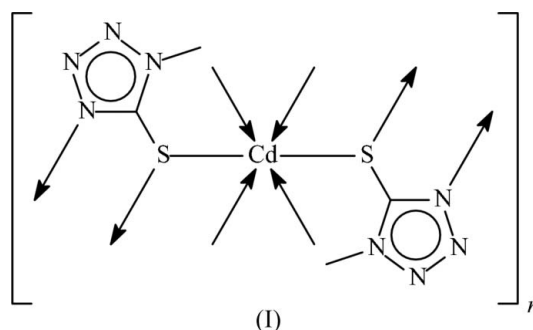
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Key indicators

Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{N}-\text{C}) = 0.005$ Å
 R factor = 0.029
 wR factor = 0.078
Data-to-parameter ratio = 15.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Poly[bis(μ -1-methyl-1,2,3,4-tetrazol-5-yl-
thiolato)cadmium(II)]The crystal structure of the title compound, $[\text{Cd}(\text{C}_2\text{H}_3\text{N}_4\text{S})_2]_n$, consists of octahedrally coordinated Cd atoms that are linked by the anionic ligands into a layer structure. The Cd atom lies on a special position of site symmetry $\bar{1}$.Received 6 November 2006
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Comment

The crystal structures of only a few metal complexes of 1-substituted 1,2,3,4-tetrazol-5-thioles have been reported; the metals comprise gold (Ortner & Abram, 1999), palladium (Kim *et al.*, 2003, 2004, 2005) and platinum (Kim *et al.*, 2003) only. The title compound, (I), adopts a layer structure in which the monoanion binds covalently through the S atom, and datively through the N atom at the 4-position; the S atom also interacts with an adjacent formula unit to generate an octahedral environment about the metal atom, which lies on an inversion centre.The synthesis of the compound was reported a long time ago (Agarwala *et al.*, 1967); the present study complements the ^{13}C NMR assignment reported earlier (Kahn & O'Brien, 1991).

Experimental

Cadmium dichloride (18 mg, 0.1 mmol), 5-mercapto-1-methyltetrazole (23 mg, 0.2 mmol) and water (12 ml) were placed in a Teflon-lined stainless steel Parr bomb. The bomb was heated at 413 K for 48 h. The bomb was then cooled to room temperature over 36 h. Colourless crystals of (I) were isolated in about 30% yield.

Crystal data

$[\text{Cd}(\text{C}_2\text{H}_3\text{N}_4\text{S})_2]$
 $M_r = 342.69$
 Monoclinic, $P2_1/c$
 $a = 11.431$ (2) Å
 $b = 5.395$ (1) Å
 $c = 7.771$ (2) Å
 $\beta = 92.00$ (3)°
 $V = 478.95$ (17) Å³

$Z = 2$
 $D_x = 2.376$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 2.69$ mm⁻¹
 $T = 295$ (2) K
 Block, colourless
 $0.16 \times 0.14 \times 0.12$ mm

Data collection

Rigaku R-AXIS RAPID IP
diffractometer
 ω scans
Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)
 $T_{\min} = 0.353$, $T_{\max} = 0.738$

4300 measured reflections
1082 independent reflections
991 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.059$
 $\theta_{\text{max}} = 27.5^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.029$
 $wR(F^2) = 0.078$
 $S = 1.09$
1082 reflections
71 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0399P)^2 + 0.0392P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.60 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\text{min}} = -0.92 \text{ e } \text{Å}^{-3}$

Table 1

Selected bond lengths (Å).

Cd1—N1 ⁱ	2.441 (2)	Cd1—S1 ⁱⁱⁱ	2.6685 (8)
Cd1—N1 ⁱⁱ	2.441 (2)	Cd1—S1 ^{iv}	2.6612 (9)
Cd1—S1	2.6612 (9)	Cd1—S1 ^v	2.6685 (8)

Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (ii) $x, -y + \frac{1}{2}, z + \frac{1}{2}$; (iii) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$; (iv) $-x + 1, -y + 1, -z + 1$; (v) $x, -y + \frac{3}{2}, z + \frac{1}{2}$.

H atoms were positioned geometrically, with C—H = 0.96 Å, and were included in the refinement in the riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. The methyl group was rotated to fit the electron density.

Data collection: *RAPID-AUTO* (Rigaku, 2004); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2006).

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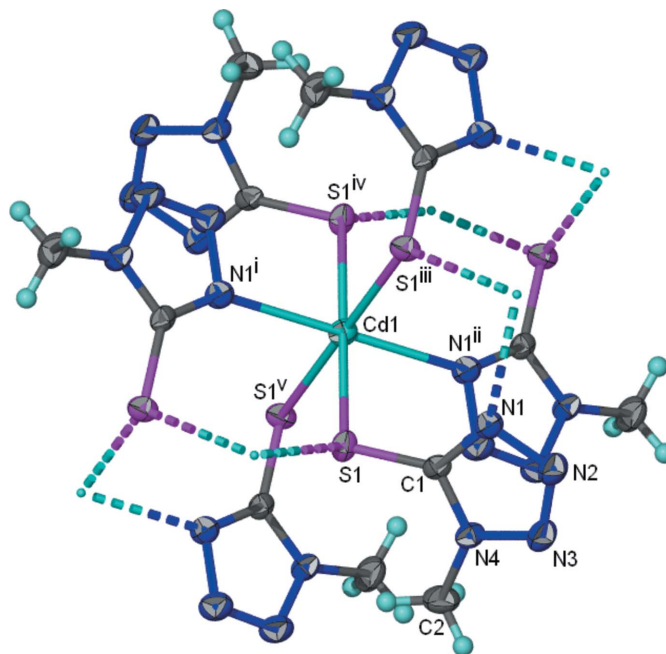


Figure 1

Displacement ellipsoid plot (70% probability) of a fragment of the layer structure of (I). H atoms are drawn as spheres of arbitrary radii. Symmetry codes are as given in Table 1. Dashed lines indicate the bridging Cd—N and Cd—S bonds.

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